

Testing a Conjecture about a Kinematics Limit in Molecular Physics



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Background

• In collisions between a diatomic molecule and a atom, a given amount of energy transfer can occur or a change to the internal energy of the molecule can occur, populating quantum states of vibrational and rotational energy. The energy available to a collision is given by:

 $E_{avail} = E_{trans} + \varepsilon(V, J)$

- We study the conjecture first published by the Valentini group (Picconatto et al. in 2001), which shows that in reactive collisions, the maximum rotational state is never populated.
 - From a theoretical perspective, if a molecule populated the maximum rotational state for the total energy, all of the translational energy in the system would be lost, and the particles would never separate.

TABLE III. $H+H_2 \rightarrow$	H_2 +H.		
Reaction (E_{trans})	$j'_{\rm max,meas.}$	$j'_{\rm max,model}$	$j'_{\rm max}(E_{\rm avail})$
$H+D_2 (1.3 \text{ eV})^a$			
v'=0	15	14	16
v' = 1	12	11	13
v'=2	8	7	9

76.4

2535 57635 763

76.25

 \neg

Table 1 (Above): A chart from the Picconatto paper depicting a collision, and the maximum populated final J state three ways: experimentally, calculated through their model, and with the maximum energy available.

Res	u	ts

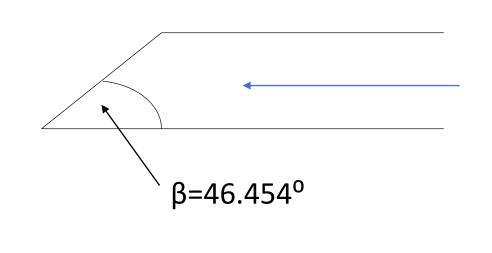
$Li_2+Xe Reaction, v_i=0, j_i=10$			
	jmax, measured	jmax, Picconatto Model	j_{max} , E_{avail}
v'=0	65.02	56.50	72.14
v'=1	63.50	51.65	68.35

Li ₂ +Xe Reaction, $v_i=0$, $j_i=20$			
	jmax, measured	jmax, Picconatto Model	j_{max} , E_{avail}
v'=0	67.16	59.34	74.52
v'=1	65.51	54.73	70.85

Li ₂ +Xe Reaction, $v_i=0$, $j_i=30$			
	jmax, measured	j _{max} , Picconatto Model	j_{max} , E_{avail}
v'=0	70.88	63.79	78.32
v'=1	68.68	59.51	74.82

Jmax, measured Jmax, Picconatto Model Jmax, E _{avail}	
v'=0 76.3639 69.5601 83.3609	
v'=1 74.2940 65.6316 80.0628	

- This work looks at non-reactive collisions such as Li₂+Xe instead, addressing a question not directly considered in the original work.
- For the Li₂+Xe collision, we analyze the effect of changing the initial energy on the maximum expected vs actually populated rotational state.
- Picconatto introduces a model for the total internal energy available where the total energy is scaled by a "skew angle", an angle in a two-dimensional representation of an atommolecule potential energy surface (shown below for the Li_2 +Xe system).



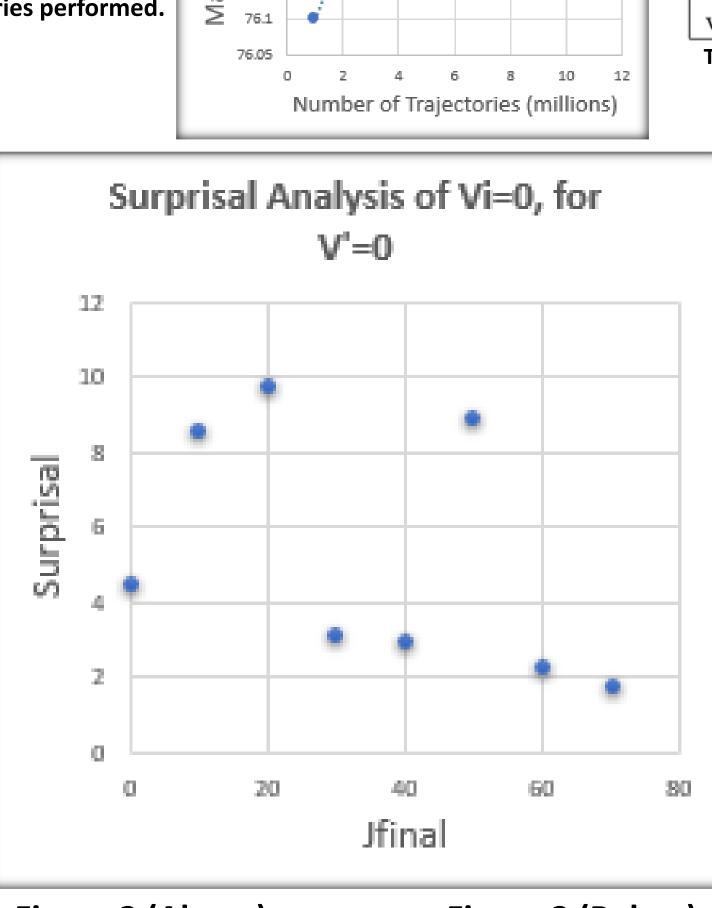
• For the Li_2 +Xe system, the skew angle is 46.454°. A collision can be represented by a line entering the surface at a given angle, and the changes in energy are shown through the reflection of the line against the walls of the box. The line bounces off of the skewed wall, and populates different j and v states.

Methods

Simulation of Trajectories

• We carry out a Monte-Carlo simulation of the collision dynamics using the fast action-angle algorithm devised by Neil Smith in order to simulate batches of trajectories based on a set of initial conditions.

Figure 1 (Right): A comparison of the maximum populated rotational state to 76.15 the number of trajectories performed.



Maximum Rotational State by

Number of Trajectories for Vi=0 Ji=40

Figure 2 (Above) Figure 3 (Below) Surprisal Analysis of Vi=0, for V'=1

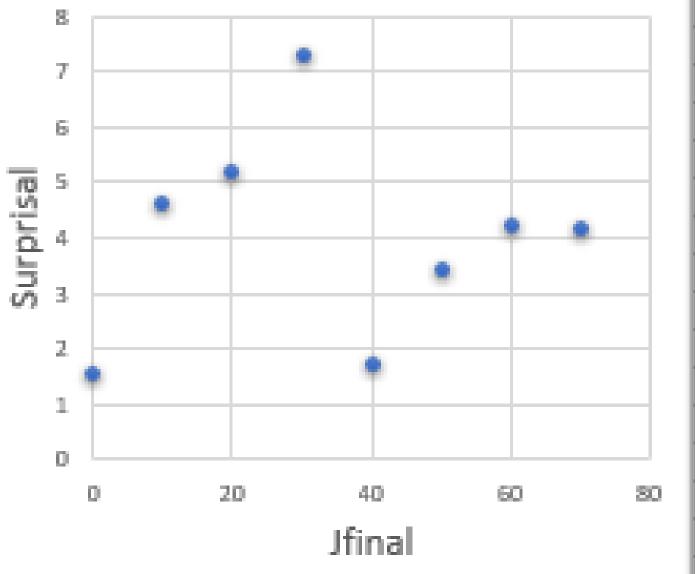


Table 2 (Above)

- **Figure 1** depicts how maximum final rotational state based on the number of trajectories simulated begins to converge at j=76.4 after 10 million trajectories.
- **Table 2** shows the maximum j values for our simulated collision. The first column represents the change in vibrational state, the second is the highest populated j state in our simulations, the third is the highest populated j state using the Picconatto model, and the fourth is the highest populated j state just considering the total energy. We format it this way to compare with the data compiled by Picconatto in **table 1**.
- Within **table 2**, we can see that the Valentini conjecture holds true-the simulated trajectories never populate states as high as the maximum state suggests that it could. However, their proposed model stays far under the maximum simulated rotational state for all of the data.
- Surprisal analysis shows how much dynamical information is contained within the experimental data, as opposed to how much information is purely statistical. To perform it, we compare our data with a model of the probability of certain events happening. In our case, we compare the probability of finding the

- We vary the relative collision speed of the system in order to vary the total energy of the system.
- We use the Peterson-McBane $Li_2 + Xe$ ab initio potential to perform these calculations, and then determine the highest-populated rotational state from our computational data.
- From there, we calculate the expected maximum rotational state based on a statistical model, along with the expected maximum rotational state based on the model proposed by Picconatto et. al...
- We are then able to perform surprisal analysis on our results to find the deviation of our experimental/simulated data from a prior statistical model.



- First, it is notable that an increase in the number of simulated trajectories does not increase the maximum populated j_{final} linearly. Instead, it begins to converge to a specific maximum j. We can reasonably say that after 10 million trajectories run, we can get the maximum j to 4 significant digits.
- Second, it seems that the model proposed by Picconatto et. al. does not hold up for non-reactive molecular collisions, at least with the system that we study. Further work will be necessary to explore whether this is true for other non-reactive systems.
- Finally, when we perform our surprisal analysis, we use the RRHO model to obtain our prior distributions. While this makes sense for the point we are at right now in the work, it will be important moving forward to create an even better model to obtain prior distributions.

maximum j experimentally with the probability of finding it using the Rigid Rotor Harmonic Oscillator model of molecular collisions.

Figures 2 and 3 each show the results of surprisal analysis being performed on different sets of data. Our surprisal analysis shows that the experimental data does not especially cohere to the model that we usethe surprisal is generally high, which indicates a large deviation from the data. However, the surprisal does not seem to follow any notable trends.

References

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